

Characterization of a novel Mg-conductor using Neutron Methods

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Neutrons are a unique probe for non-destructive structural studies of energy materials, especially for future investigation and development of highly conductive solid state Mg electrolyte, neutrons are one of the main requirements for successful post-Li battery research. Recently, a new compound synthesized from γ -Mg(BH₄)₂ and ethylenediamine (C₂H₈N₂, abbreviation “en”) was reported to have an exceptionally high magnesium ion conductivity of up to $6 \times 10^{-5} \text{ S cm}^{-1}$ at 70 °C in the solid state. In our work, the structure of this new compound has been solved and shows a different ratio of the precursors, γ -Mg(BH₄)₂ : [Mg(en)₃(BH₄)₂], while initially reported was 2:1. A new ratio of precursors will increase the ionic conductivity, simply because there is less unreacted γ -Mg(BH₄)₂. High resolution neutron powder diffraction data was previously collected at the NOVA beamline at J-Parc Spallation Source, Japan, and shows a very good correlation with the proposed model. Conductivity measurements will be presented as well as quasi elastic neutron scattering (QENS) experiments.

Reference

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