

Effect of hydrostatic pressure on the stability of mono- and multi-molecular G-Quadruplexes

Y. Li, U.N. Feroze, N. Tariq, R. Abu-Ghazalah^a, A. Amiri, R.B. Macgregor, Jr.
Graduate Department of Pharmaceutical Sciences, Leslie Dan Faculty of Pharmacy, University
of Toronto M5S 3M2 Canada; ^aW. Booth School of Engineering Technology Practice and
Technology, McMaster University, L8S 0A3 Canada

Using spectroscopy, we have studied the effect of hydrostatic pressure on the conformational stability of monomolecular and multimolecular 4-stranded DNA molecules (G4). In earlier work, we showed that pressure destabilizes the G4 structure formed by the monomolecular, single stranded oligonucleotide, HTel¹. The folded structure of HTel has three G-tetrads and three loops, that remain single stranded. The molar volume change of unfolding HTel, ΔV , is approx. $-50 - -60 \text{ cm}^3 \text{ mol}^{-1}$ at $T_m \sim 60 \text{ }^\circ\text{C}$. Thus, it is de-stabilized by pressure. Mechanistically, the large, negative volume change was assumed to arise predominately from the release of the cations coordinated by the G-tetrads. In an effort to better understand the behaviour of G4 molecules at elevated pressure, we have studied several oligonucleotides with sequences similar to that of HTel and the tetramolecular G4, $d(\text{TG}_4\text{T})_4$, a four-stranded molecule without loops. The molar volume changes observed for the modified HTel oligo-nucleotides are all negative, they are all destabilized by pressure. By looking at the effect of pressure on $d(\text{TG}_4\text{T})_4$, we hoped to obtain information on the volume change that arises from the destabilization of the G-tetrad. However, the pressure does not appear to influence the stability of $d(\text{TG}_4\text{T})_4$; thus, $\Delta V \sim 0 \text{ cm}^3 \text{ mol}^{-1}$ under the conditions of our study. This unexpected result implies that the factors that contribute to the volume change arising from the formation of tetra-molecular GQ structures balance each other with the result that the net ΔV is zero.

¹ J. Am. Chem. Soc. 2011, 133, 4518–4526